



Course Specification

(Bachelor)

Course Title: **Stereochemistry**

Course Code: **CEM 333**

Program: **Chemistry**

Department: **Chemistry**

College: **Science**

Institution: **Majmaah University**

Version: **TP -153**

Last Revision Date: **15 December 2024**



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A. General information about the course:

1. Course Identification

1. Credit hours: (2)

2. Course type

A. University College Department Track Others
 B. Required Elective

3. Level/year at which this course is offered: (5th level- third year)

4. Course General Description:

Stereochemistry the branch of chemistry concerned with the three dimensional aspects of molecules. The chemical studies take into account the spatial aspects of molecules.

Stereochemistry course introduce the fundamental principles in all aspects of stereochemistry. This concise resource is generously enriched with numerous worked examples.

The course provide definitions of terms such as chirality, enantiomers, diastereoisomers and racemisation, complete with suitable examples to illustrate key concepts. The use of a polarimeter and associated definitions ar described, together with two different conventions D, L and R, S for specification of configuration. The distinction between conformation and configuration

This course explains how the different properties of stereoisomers of a compound arise, and what processes can be used to prepare and analyze stereoisomerically pure compounds.

5. Pre-requirements for this course (if any):

CEM231

6. Co-requisites for this course (if any):

None

7. Course Main Objective(s):

- Explore the field of stereochemistry, including nomenclature, analytical techniques, asymmetric syntheses and real-world applications.
- Provide fundamental principles in all aspects of stereochemistry. This concise resource is generously enriched with numerous worked examples.
- Explains how the different properties of stereoisomers of a compound arise, and what processes can be used to prepare and analyze



- Apply various concepts of stereochemistry and fundamental principles of stereoselectivity in organic chemistry

2. Teaching mode (mark all that apply)

No	Mode of Instruction	Contact Hours	Percentage
1	Traditional classroom	30	100%
2	E-learning		-
3	Hybrid <ul style="list-style-type: none"> Traditional classroom E-learning 		-
4	Distance learning		-

3. Contact Hours (based on the academic semester)

No	Activity	Contact Hours
1.	Lectures	30
2.	Laboratory/Studio	-
3.	Field	-
4.	Tutorial	-
5.	Others (specify)	-
Total		

B. Course Learning Outcomes (CLOs), Teaching Strategies and Assessment Methods

Code	Course Learning Outcomes	Code of PLOs aligned with the program	Teaching Strategies	Assessment Methods
1.0	Knowledge and understanding			
1.1	Recognize the stereochemistry of organic compounds, and the different types of isomerism	K1	Formal lectures, Effective Learning Collaborative Learning	Standardized exam Laboratory examination - Class exercises



Code	Course Learning Outcomes	Code of PLOs aligned with the program	Teaching Strategies	Assessment Methods
				- Evaluation of research
2.0	Skills			
2.1	Communicate effectively orally and written using appropriate presentation formats for different issues with recipients of different types	S2	Formal lectures, Effective Learning Collaborative Learning Team-based learning	Standardized exam Laboratory examination - Class exercises - Evaluation of research
2.2	Design the organic isomer effective Stereoisomers and conformational Stereoisomers using models and technology	S3		
3.0	Values, autonomy, and responsibility			
3.1	Demonstrate the ability of working independently and with groups	V3	Interactive learning Cooperative learning	Performance appraisal Behavioral observation Assessment of Reports Assessment of Group presentations Indirect assessment

C. Course Content

No	List of Topics	Contact Hours
1.	Introduction to stereochemistry of organic compounds, Different types of isomerism in organic compounds, Isomerism: Constitutional Isomers	2





	and Stereoisomers, Chirality and Stereochemistry, Importance of stereochemistry in real life: some examples	
2.	Constitution and Configuration; Chemistry in 3D space Chirality and its origin, Molecules Having One Chirality Center are Chiral symmetry criterion	2
3.	Enantiomers and Chiral Molecules Structure of Enantiomers Nomenclature for Enantiomers (Speciation of Absolute Configuration) R/S Nomenclature system (Cahn–Ingold–Prelog convention) Chemical Properties of Enantiomers, Physical Properties of Enantiomers	4
4.	Properties of Enantiomers: Optical activity, the origin of optical activity, Polarimeter, Clockwise Dextrorotatory (+), Counterclockwise Levorotatory (-), Enantiomers can Plane Polarized Light Racemization, Formation of racemic mixtures, Enantiomeric Excess, Methods for determining enantiomeric excess, The Biological Importance of Enantiomers	4
5.	Molecules with More than One Chirality Center, Number of Stereoisomers, Meso Compound.	2
6.	Diastereomers, Properties of Diastereomers, Geometric isomerism, spatial selectivity in certain organic Reactions, Stereoisomerism of Cyclic Compounds, Identification of Asymmetric Carbons in Cyclic Compounds.	4
7.	Fisher and Newman projection of stereo compounds, Fischer Projection Formulas, Assigning R,S Configuration to Fischer Projections, Naming from the Fischer Projection Draw the stereocompounds in 2D, Newmann and the conformation of alkane	4
8.	Stereoelectronic and steric principles in reactions: Substitution, elimination and addition; selectivity and specificity.	4
9.	Conformations of acyclic molecules conformational isomerism, Ethane, Propane, Butane conformations, structural. Conformations of cyclic alkanes and their derivatives Stereoisomers, conformers, conformations of cyclohexane Rings, Cyclohexane The chair conformation, The boat conformation, The twist boat conformation, Interconversion of cyclohexane conformations.	4

Total		30





D. Students Assessment Activities

No	Assessment Activities *	Assessment timing (in week no)	Percentage of Total Assessment Score
1.	Homework, exercises, periodic tests , Essays, project , Presentation	During the semester	20%
2.	Mid term 1	6 th week	15%
3.	Mid term 2	11 th week	15%
4.	Electronic exam	12 th week	10%
5.	Final exam	End of semester	40%

*Assessment Activities (i.e., Written test, oral test, oral presentation, group project, essay, etc.).

E. Learning Resources and Facilities

1. References and Learning Resources

Essential References	Ranjit S. Dhillon 2014. Stereochemistry, Alpha, Science Intl Ltd
Supportive References	<u>T. W. Graham Solomons</u> , <u>Craig B. Fryhle</u> , <u>Scott A. Snyder</u> . Organic Chemistry 12th Edition 2017.
Electronic Materials	https://www.khanacademy.org/science/organic-chemistry/stereochemistry-topic https://www.researchgate.net/publication/261991150_Stereochemistry
Other Learning Materials	ChemDraw Professional 17.0 Suite ACD/ChemSketch :: Draw Chemical Structures :: ACD/Labs https://chemaxon.com/marvin

2. Required Facilities and equipment

Items	Resources
facilities (Classrooms, laboratories, exhibition rooms, simulation rooms, etc.)	Class room, laboratory of Organic chemistry
Technology equipment (projector, smart board, software)	The electronic platform, data show, Smart Board
Other equipment (depending on the nature of the specialty)	



F. Assessment of Course Quality

Assessment Areas/Issues	Assessor	Assessment Methods
Effectiveness of teaching	Students	Student evaluation (electronically questionnaire) organized by the University
Effectiveness of Students assessment	Department	Analysis of electronically questionnaire. the Make decision through department Council
Quality of learning resources	Department / staff members	Analysis of course report by Chemistry Department Council
The extent to which CLOs have been achieved	Department Faculty	CLO Analysis Report
Other		

Assessors (Students, Faculty, Program Leaders, Peer Reviewers, Others (specify))

Assessment Methods (Direct, Indirect)

G. Specification Approval

COUNCIL /COMMITTEE	Chemistry Department Council	
REFERENCE NO.	1	
DATE	14/6/1446 H	15/12/2024

