

## معلومات المقرر \* (Course Information):

	كيمياء ضوئية	اسم المقرر:
	CHM336	رقم المقرر:
	اختباري	اسم ورقم المتطلب السابق:
	لا يوجد	اسم ورقم المتطلب المرافق:
	السادس	مستوى المقرر:
	٧ .	الساعات المعتمدة:
Photochemistry		
CHEM336		
Elective		
None		
₹th level		
Υ		
	CHEM336  Elective  None  'th level	CHM336  اختباري لا يوجد السادس  Photochemistry  CHEM336  Elective  None

**Module Description** 

وصف المقرر:

The course covered: Basic rules in photochemistry, definition of light, photon.

The laws of photochemistry: Laws of Gruthus, Einstein ,Beer Lambert and Perre. Electronic transitions, absorption of light, Potential Energy Curve. Franck–Condon principle. jablonski energy diagram Dissolution of excited state, equation of dissolution of excited state

Fluorescence, Phosphorescence .Photochemical reaction.

Laser: type of Laser, Danger of Laser. Sun radiation. Applications of sun energy,

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Department of Chemistry

College of Science

in Zuffi





# أهداف المقرر :

1	Studying the light, photon, Electromagnetic Radiation.	1
2	Studying the laws of photochemistry	۲
3	Studying electronic transitions, absorption of light,	٣
4	Studying potential energy curve. Franck-Condon principle	٤
5	Studying dissolution of excited state, equation of dissolution of excited state	٥
6	Studying fluorescence, Phosphorescence,	٦
7	Studying laser, sun radiation.	Y

## **Learning Outcomes:**

مخرجات التعليم:

### successful students will be able to:

1	Define the light and photon nature	١
2	Discuss laws of photochemistry	۲
3	Knowledge of the electronic transitions, absorption of light	٣
4	Explain potential energy curve	٤
5	Knowledge of dissolution of excited state, equation of dissolution of excited state	0
6	Differentiate between fluorescence, Phosphorescence	
7	Compare between types of laser	
8	Knowledge of sun energy .	M
	Department of Scient	



10.00 · 10.00

M C CA S A SA



#### **Course Contents:**

محتوى المقرر:

ساعات التدريس	عدد الأسابيع	قائمة الموضوعات		
(Hours)	(Weeks)	(Subjects)		
4	۲	Basic rules in photochemistry, definition of light, photon.  The laws of photochemistry: Laws of Gruthus, Einstein ,Beer Lambert and Perre.		
٤	۲			
٤	۲	Electronic transitions, absorption of light		
٤	۲	Potential Energy Curve. Franck–Condon principle.  jablonski energy diagram		
٤	۲	Dissolution of excited state, equation of dissolution of excited state		
۲	١	Fluorescence, Phosphorescence		
۲	١	Photochemical reaction		
۲	١	Danger of Laser: type of Laser,		
٤	۲	Sun radiation. Applications of sun energy		

#### **Textbook and References:**

الكتاب المقرر والمراجع المساندة:

ISBN	سنة النشر	اسم الناشر	اسم المؤلف (رئيسي)	اسم الكتاب المقرر
IJDIN		Publisher		
	Publishing Year		Author's Name	Textbook title
174.571577	79	A John Wiley & Sons	Brian Wardle	
				Principles and
				Applications
				of Photochemistry
	سنة النشر	اسم القاشر	اسم المؤلف (رئيسي)	اسم المرجع
	Publishing Year	Publisher	<b>Author's Name</b>	Reference
. 7 7 7 1 9 5 7 7	79		Yuri V. Il'ichev	
		Johnson and		PHOTOCHEMISTRY
		Johnson Company		Theoretical Concepts
				and Reaction
				Mechanisms
	۲۰۰۸		Nicholas J. Turro, J. C	Principles of Molecula
				Photochelstry:An
				College of Science in Zulfi

صفحة ٣ من 3





\*يتم تعبئة معلومات المقرر فقط باللغتين العربية والانجليزية وباقي المعلومات بلغة التدريس المعتمدة ويكرر لكل مقرر في الخطة الدراسية

\* Course Information should be filled in Arabic and English. Other information should be filled using the approved teaching language at the college.



