

## معلومات المقرر \* (Course Information):

	كيمياء عامة- ٢	اسم المقرر:
	CHM102	رقم المقرر:
	کیمیاء عامة- ۱، CHM101	اسم ورقم المنطلب السابق:
	لا يوجد	اسم ورقم المنطلب المرافق:
	المستوي الثاتي	مستوى المقرر:
	(۲+۲)۳	الساعات المعتمدة:
Module Title:	General Chemistry- 2	
Module ID:	CHM102	
Prerequisite (Co-requisite):	General chemistry-1, CHM-101	
Co-requisite:	None	
Course Level:	Second Level	
Credit Hours:	3(2+2)	

وصف المقرر:

This course includes:

Principle of atomic structure (Bohr's theory , electromagnetic radiation, regions of electromagnetic radiations- wave length for each region). Quantum numbers, Electronic configurations for many electronic atoms- aufbau principle-Pauli exclusion principle- Hund's rule). Periodic table and Periodicity of properties with atomic number- ionization energy- electron affinity-electro negativity. Chemical bonding (ionic and covalent) Lewis structure, hybridization, resonance, introduction to valence bond theory. chemical kinetics, rate law and the relation between its variables.

أهداف المقرر:

- Studying the Principle of atomic structure (Bohr's theory, electromagnetic radiation, region electromagnetic radiations- wave length for each region)
- 2 Studying the quantum numbers, Electronic configurations for many electronic atoms raufbat principle-Pauli exclusion principle- Hund's rule

صفحة ١ من 3



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3	Studying the periodic table and periodicity of properties with atomic number( ionization energy- electron affinity-electro negativity)	٣
4	Studying ionic bond and the factors that affect the ionic bonds. Lewis structure, covalent bonding, hybridization, resonance, introduction to valence bond theory	٤
5	Studying the chemical kinetics, rate law and the relation between its variables.	٥

مخرجات التعليم:

## Learning Outcomes: successful students will be able to

1	Demonstrate knowledge and understanding of the basics and concepts in chemistry	,
2	Describe the electron configuration of the elements.	۲
3	Apply the knowledge and skill to solve problems relating to course concept	
4	Explain the fundament of the basic chemistry.	٣
5	Work independently and as part of a team	
6	Able to use a computer as a tool in writing, drawing chemical structures and Saudi Digital Library.	
7	Follow the safety precautions in the laboratory and perform the experiments in lab.  Prepare standard solutions from solid or liquid materials  Perform simple titration experiments in neutralization reactions	٦

## **Course Contents:**

محتوى المقرر:

ساعات التدريس	عدد الأسابيع	قائمة الموضوعات		
(Hours)	(Weeks)	(Subjects)		
4	2	atomic structure		
6	3	quantum numbers and electronic configuration of different elements		
4	2	the periodic table and periodicity of properties with atomic number( ionization energy- electron affinity-electro negativity		
4	2	Chemical bonding		
4	2	Hybridization of atomic orbitals (sp³, sp², and sp) and molecular shapes – introduction to valence bond theory		
4	2	Intermolecular interactions (van der Waals equation, intermolecular forces)		

صفحة ٢ من 3





2	1	Acid-Base theories	
2	1	Chemical Kinetics	
		Practical experiments for theoretical lectures.	
2	1	Safety in chemical lab and identification of apparatus and tools.	
4	2	Preparation of standard and molar solutions	
6	3	Preparation of standard solutions by dilution	
6	3	Simple titrations	
2	1	Determination of density	
4	2	Determination of viscosity	
2	1	Revision	

## **Textbook and References:**

الكتاب المقرر والمراجع المسائدة:

ISBN	سنة النشر	اسم الناشر	اسم المؤلف (رئيسي)	اسم الكتاب المقرر
	<b>Publishing Year</b>	Publisher	<b>Author's Name</b>	Textbook title
978471528746	2000	John Wiley & Sons, Inc	James E. Brady	General Chemistry: Principles and Structure
ISBN	سنة النشر	اسم الناشر	اسم المؤلف (رنيسي)	اسم المرجع
	<b>Publishing Year</b>	Publisher	<b>Author's Name</b>	Reference
978-007-127220-9	2010	McGraw Hill	Raymond Chang	Chemistry 10th edition
13: 978-1-429- 21812-2 1-4292-1812-6	2010	Freeman Publishers	Peter Atkins and Julio De Paula	Physical Chemistry
100078021510 139780078021510	2015	McGraw Hill	Raymond Change and Kenneth Goldsby	General Chemistry: The Essential concept

"يتم تعبئة معلومات المقرر فقط باللغتين العربية والانجليزية وباقي المعلومات بلغة التدريس المعتمدة ويكرر لكل مقرر في الخطة الدراسية

\* Course Information should be filled in Arabic and English. Other information should be filled using the approved teaching language at the college.



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